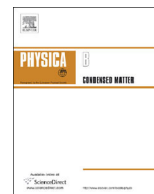


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# Two-photon, three-photon, and four-photon excellent near-infrared quantum cutting luminescence of $\text{Tm}^{3+}$ ion activator emerged in $\text{Tm}^{3+}:\text{YNbO}_4$ powder phosphor one material simultaneously

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## ABSTRACT

In present study, two-photon, three-photon, and four photon near-infrared quantum cutting luminescence of  $\text{Tm}^{3+}$  ion activator in  $\text{YNbO}_4$  powder phosphor is reported. The visible to near-infrared excitation and emission spectra and fluorescence lifetimes of  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  powder phosphor are measured.  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  is found to possess intense two-photon, strong three-photon, and moderate four-photon quantum cutting 1820 nm  $^3\text{F}_4 \rightarrow ^3\text{H}_6$  luminescence of the  $\text{Tm}^{3+}$  ion simultaneously. The up-limit of the two-, three-, and four-photon near-infrared quantum cutting efficiency are found to be approximately 166%, 198%, and 192%, respectively. These results are expected to be valuable in aiding the probing of new generation environmentally friendly germanium Ge solar cells, currently a popular condensed matter physical topic globally.

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## 1. Introduction

Sustainable energy production based on the direct conversion of solar energy is becoming increasingly important as it may be the only source capable of supplying sufficient energy for long-term worldwide energy requirements [1,2]. Quantum cutting materials with quantum efficiency obviously higher than 100% have attracted great interest for their potential applications in solar cells [1–12]. Since the visible quantum cutting luminescence phenomena in  $\text{Eu}^{3+}\text{--}\text{Gd}^{3+}$  material was reported by Meijerink et al. in Science [3], the importance, application and significance of the quantum cutting have been widely recognized [1–25]. Although the solar spectrum ranges from 280 nm to 2500 nm at AM 1.5 G [15], the efficiency of normal single junction silicon solar cells is limited to 30% of the Shockley–Queisser limit. More than 70% loss is caused by spectral mismatch losses of transmission and thermalization loss. In 2002, Green et al. proposed the two-photon quantum cutting solar cell theory for the first time, and found that

the maximum efficiency of two-photon quantum cutting silicon solar cells can reach 39.6% [10], which are sensitive to the wavelength range 280–1100 nm. Meijerink et al. reported the first experiment phenomena work for second-order near-infrared quantum cutting luminescence in  $\text{Yb}_x\text{Y}_{1-x}\text{PO}_4:\text{Tb}^{3+}$  phosphor in 2005 [1]. From 2007, Meijerink [1,15,21], Qiu and Zhou [17], Wang and Chen [5,6], Huang, and Zhang [7], etc. [9,11,23] have reported about 200 literatures of experimental researches of two-photon second-order near-infrared quantum cutting of Sentiizer- $\text{Yb}^{3+}$  ion codoped materials, which is used to develop two-photon quantum cutting silicon solar cells [16,19,20]. Qiu and Zhou [2,12], Zhang and Huang [14], our group [8,25] have reported experimental researches of multi-photon first-order near-infrared quantum cutting of  $\text{Tm}^{3+}$  or  $\text{Er}^{3+}$  ion activator doped materials from 2009. This further improvement is the multi-photon quantum cutting germanium Ge or silicon-germanium Si–Ge solar cell [2,8,12,14,25], which is sensitive to wavelengths of 280–1850 nm or 280 to about 1650 nm and is environmentally friendly. Its maximum efficiency is able to greatly exceed 39.6% [22] using multi-photon infrared quantum cutting to largely reduce transmission and thermalization losses because the energy band-gap,

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$E_g$  of Ge is positioned at about 1850 nm (0.67 eV:300 K) [2,8,12,14,25]. Above works are very significant. The present study reports the two-photon, three-photon, and four-photon near-infrared quantum cutting luminescence phenomena in  $\text{Tm}^{3+}$  ion doped  $\text{YNbO}_4$  phosphor material. To the knowledge of the authors, this is the first time to find that the effective two-photon, three-photon, and four-photon near infrared quantum cutting are emerged in one material simultaneously. The  $\text{Tm}^{3+}$  ion is a new attractive near-infrared quantum cutting activator because its quantum cutting process is first-order. This property is useful because, as Meijerink et al. have pointed out [2,8,12,14,25], first-order energy transfer mechanisms generally have a much higher probability (typically a factor of 1000) of occurring than second-order mechanisms. However, so far there have only been ten reports regarding the near-infrared quantum cutting of the about 1800 nm luminescence of the  $\text{Tm}^{3+}$  activator center [8,14]. Therefore, more thorough study of multi-photon infrared quantum cutting luminescence of  $\text{Tm}^{3+}$  ion activator is urgently required.

## 2. Experiment instrument and the sample

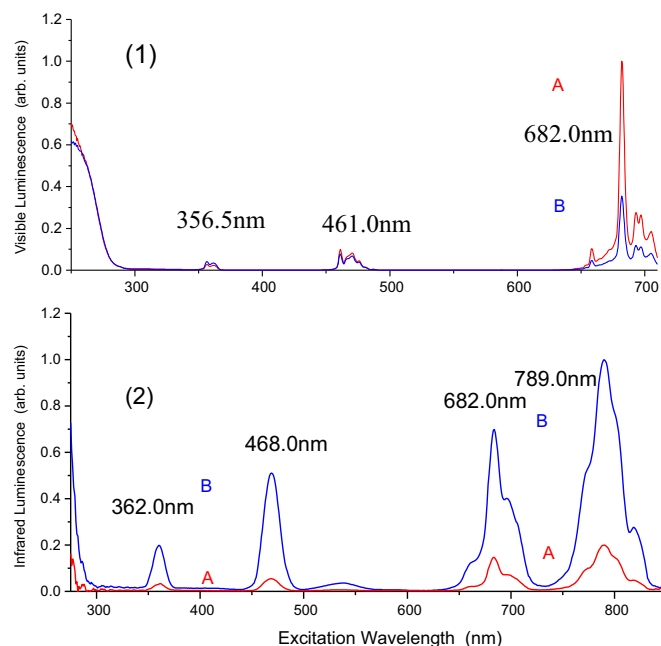
The samples used in our experiment were (A)  $\text{Tm}_{0.005}\text{Y}_{0.995}\text{NbO}_4$  and (B)  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  powder phosphors, which were prepared through a solid-state reaction. The starting materials were high-purity  $\text{Y}_2\text{O}_3$  (99.99%),  $\text{Nb}_2\text{O}_5$  (99.99%), and  $\text{Tm}_2\text{O}_3$  (99.99%). The raw powders were weighed according to the stoichiometric compositions and then mixed, ground, and pressed into pellets. Finally, the powders were sintered at 1400 °C for 5 h in a furnace in air. A FL3-2iHR fluorescence spectrometer (Horiba-JY Co., America, Japan, and France) was used to investigate the luminescence of the phosphor powders.

## 3. The excitation spectra and luminescence spectra and lifetime

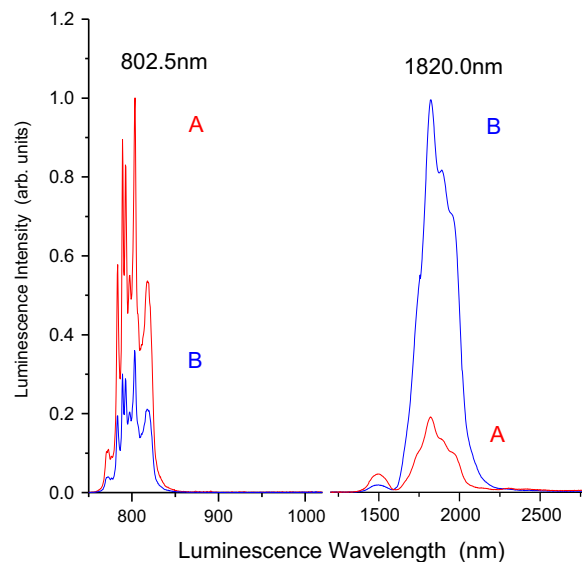
First, the visible excitation spectra of 802.5 nm received wavelength of (A)  $\text{Tm}_{0.005}\text{Y}_{0.995}\text{NbO}_4$  and (B)  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  powder phosphors in the wavelength range of 250–710 nm were measured, as shown in part (1) of Fig. 1. The  $\text{Tm}^{3+}:\text{YNbO}_4$  powder phosphor was found to exhibit three groups of excitation spectra signal peaks from ultraviolet to visible. It is found that their main peaks were positioned at 356.5, 461.0 and 682.5 nm. These are easily recognizable as the  $^3\text{H}_6 \rightarrow ^1\text{D}_2$ ,  $^3\text{H}_6 \rightarrow ^1\text{G}_4$ , and  $^3\text{H}_6 \rightarrow ^3\text{F}_3$  absorption transitions, respectively, of the  $\text{Tm}^{3+}$  ion [12–18].

The excitation spectra of 1820.0 nm wavelength of (A)  $\text{Tm}_{0.005}\text{Y}_{0.995}\text{NbO}_4$  and (B)  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  powder phosphor in the wavelength range of 275–850 nm are shown in part (2) of Fig. 1. The  $\text{Tm}^{3+}:\text{YNbO}_4$  powder phosphor exhibited four groups of excitation spectra signal peaks between 275 and 850 nm. It is found that their main peaks were positioned at 362.0, 468.0, 682.5, and 789.0 nm. These clearly correspond to the  $^3\text{H}_6 \rightarrow ^1\text{D}_2$ ,  $^3\text{H}_6 \rightarrow ^1\text{G}_4$ ,  $^3\text{H}_6 \rightarrow ^3\text{F}_3$ , and  $^3\text{H}_6 \rightarrow ^3\text{H}_4$  absorption transitions, respectively, of the  $\text{Tm}^{3+}$  ion [12–18].

The visible luminescence spectra of the (A)  $\text{Tm}_{0.005}\text{Y}_{0.995}\text{NbO}_4$  and (B)  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  powder phosphors were then measured. The luminescence spectra at 750–1020 nm, when  $^3\text{F}_3$  state of the  $\text{Tm}^{3+}:\text{YNbO}_4$  powder phosphor is excited by 682.5 nm light, were measured, as shown in Fig. 2. A medium-sized luminescence peak at (789.0, 802.5 nm) assigned to the  $^3\text{H}_4 \rightarrow ^3\text{H}_6$  fluorescence for the  $\text{Tm}^{3+}:\text{YNbO}_4$  powder phosphor was observed. The luminescence spectra at 510–850 nm, when  $^1\text{G}_4$  state of the  $\text{Tm}^{3+}:\text{YNbO}_4$  powder phosphor is excited by 461.0 nm light, were measured, as shown in Fig. 3. In this case, two medium-sized luminescence peaks at 648.0 nm and (791.0, 802.5 nm) of the  $^1\text{G}_4 \rightarrow ^3\text{F}_4$  and  $^3\text{H}_4 \rightarrow ^3\text{H}_6$  fluorescence of the  $\text{Tm}^{3+}:\text{YNbO}_4$  powder



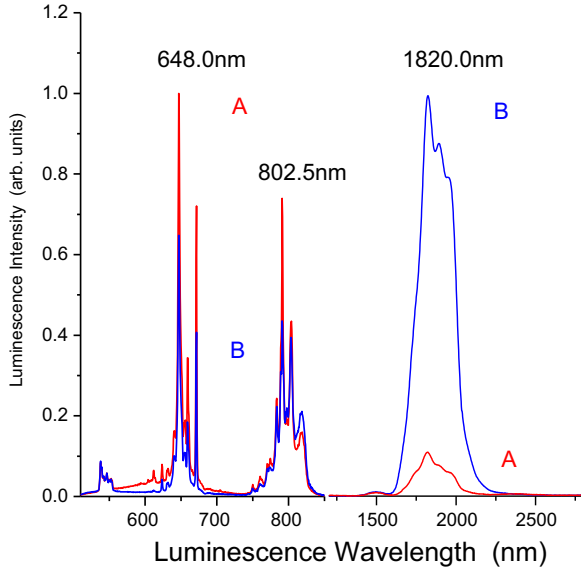
**Fig. 1.** Visible (1) and infrared (2) excitation spectra of (A)  $\text{Tm}_{0.005}\text{Y}_{0.995}\text{NbO}_4$  (red) and (B)  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  (blue) powder phosphors obtained when the fluorescence received wavelength is positioned at 802.5 nm (1) and 1820.0 nm (2). (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)



**Fig. 2.** Luminescence spectra of (A)  $\text{Tm}_{0.005}\text{Y}_{0.995}\text{NbO}_4$  (red) and (B)  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  (blue) powder phosphor obtained when the 682.5 nm  $^3\text{H}_6 \rightarrow ^3\text{F}_3$  excitation peak was selected as the excitation wavelength. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

phosphor were observed. The luminescence spectra at 418–700 and 700–1000 nm, when  $^1\text{D}_2$  state is excited by 356.5 nm light, were measured. A strong luminescence peak at (452.5, 456.0 nm), a small peak at (656.5, 662.0, 666.5 nm), and a medium-sized peak at (789.0, 802.5 nm) were observed, corresponding to the luminescence transitions of  $^1\text{D}_2 \rightarrow ^3\text{F}_4$ ,  $^1\text{G}_4 \rightarrow ^3\text{F}_4$ , and  $^3\text{H}_4 \rightarrow ^3\text{H}_6$ , respectively [12–18]. The luminescence spectra when excited by 356.5 nm are similar to that when excited by 682.5 nm and 461.0 nm light.

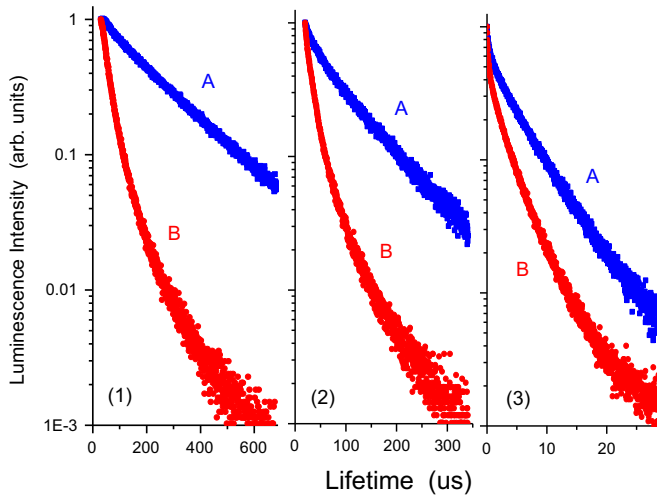
The infrared luminescence spectra of the (A)  $\text{Tm}_{0.005}\text{Y}_{0.995}\text{NbO}_4$  and (B)  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  powder phosphors were also measured.



**Fig. 3.** Luminescence spectra of (A)  $\text{Tm}_{0.005}\text{Y}_{0.995}\text{NbO}_4$  (red) and (B)  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  (blue) powder phosphor obtained when the 461.0 nm (for visible) or 468.0 nm (for infrared)  $^3\text{H}_6 \rightarrow ^1\text{G}_4$  excitation peak was selected as the excitation wavelength. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

In the same manner, the  $^3\text{F}_3$ ,  $^1\text{G}_4$ , and  $^1\text{D}_2$  state of the  $\text{Tm}^{3+}:\text{YNbO}_4$  powder phosphor were also excited by the 682.5, 468.0, and 362.0 nm light to measure the luminescence spectra at 1200–2800 nm, as shown in Figs. 2 and 3. Two groups of luminescence peaks positioned at 1496.0 nm and 1820.0 nm were observed, which are clearly the  $^3\text{H}_4 \rightarrow ^3\text{F}_4$  and  $^3\text{F}_4 \rightarrow ^3\text{H}_6$  luminescence transitions [12–18]. It can be seen that the 1820.0 nm  $^3\text{F}_4 \rightarrow ^3\text{H}_6$  near-infrared luminescence signal intensity of the (B)  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  powder phosphor is much intense than that of (A)  $\text{Tm}_{0.005}\text{Y}_{0.995}\text{NbO}_4$ .

The fluorescence lifetimes of the 802.5, 648.0, and 456.0 nm visible fluorescences of the (A)  $\text{Tm}_{0.005}\text{Y}_{0.995}\text{NbO}_4$  and (B)  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  powder phosphors, as shown in Fig. 4, were measured also. The pumping sources for the measurements were a



**Fig. 4.** Fluorescence lifetimes of the (1) 802.5 nm (left), (2) 648.0 nm (middle), and (3) 456.0 nm (right) visible luminescences of (A)  $\text{Tm}_{0.005}\text{Y}_{0.995}\text{NbO}_4$  (blue) and (B)  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  (red) powder phosphors, when excited by (1) 682.5 nm (left), (2) 461.0 nm (middle), and (3) 368.0 nm (right) pulsed light. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

360 nm NanoLED semiconductor quasi-laser and pulse Xe lamp. All the fluorescence lifetime values were fitted from the measured fluorescence lifetime experimental curves [1–2,5–8,12,14,15,16,17,19–20,25], using the tail-fit method. The origin points of fit were set after the fluorescence-spectrometer equipment response. From the analysis of measured experimental lifetime curves results, the fluorescence lifetime values of the 802.5, 648.0, and 456.0 nm fluorescences of (A)  $\text{Tm}_{0.005}\text{Y}_{0.995}\text{NbO}_4$  were found to be  $\tau_A(802.5 \text{ nm}) = 230.10 \mu\text{s}$ ,  $\tau_A(648.0 \text{ nm}) = 102.68 \mu\text{s}$ , and  $\tau_A(456.0 \text{ nm}) = 5.7760 \mu\text{s}$ , respectively; those of (B)  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  were found to be  $\tau_B(802.5 \text{ nm}) = 79.487 \mu\text{s}$ ;  $\tau_B(648.0 \text{ nm}) = 41.612 \mu\text{s}$  and  $\tau_B(456.0 \text{ nm}) = 3.4850 \mu\text{s}$ , respectively.

#### 4. Analysis

It is well-known that the total decay rate  $W_{\text{tot}}$  of  $\text{Tm}^{3+}$  ion at 0.5% concentration can be written as following [2]:

$$W_{\text{tot Tm}(0.5\%)} = A + W_{\text{MP}} = \tau^{-1}_{\text{Tm}(0.5\%)} \quad (1)$$

where  $A$  is the spontaneous radiative rate,  $W_{\text{MP}}$  is the multi-photon relaxation rate,  $\tau_{\text{Tm}(0.5\%)}$  is the luminescence lifetime of the  $\text{Tm}^{3+}$  ion whose concentration is of 0.5%. In formula (1), It is assumed that the energy transfer among the  $\text{Tm}^{3+}$  ions is very small when  $x=0.5\%$ , which can represent the case of non-energy transfer. Therefore we have that the total decay rate of  $\text{Tm}^{3+}$  ion in  $x\%$  concentration can be given by [2]:

$$W_{\text{tot Tm}(x\%)} = A + W_{\text{MP}} + W_{\text{ET}} = \tau^{-1}_{\text{Tm}(x\%)} \quad (2)$$

where  $W_{\text{ET}}$  is the energy transfer rate, and  $\tau_{\text{Tm}(x\%)}$  is the luminescence lifetime of the  $\text{Tm}^{3+}$  ion in  $x\%$  concentration. From formulae (1) and (2), we have that the energy transfer efficiency can be given by [2]:

$$\eta_{\text{ET},x\%\text{Tm}} = \frac{W_{\text{ET}}}{A + W_{\text{MP}} + W_{\text{ET}}} = 1 - \frac{\tau_{\text{Tm}(x\%)}}{\tau_{\text{Tm}(0.5\%)}} \quad (3)$$

Meanwhile, according to well-known infrared quantum cutting literature, the energy transfer efficiency of the  $\text{Tm}^{3+}$  ion caused by cross-relaxation energy transfer can be given by formula (4) also [1–2,5–8,12,14–17,19–20,25]:

$$\eta_{\text{tr},x\%\text{Tm}} \approx 1 - \frac{\int I_{x\%\text{Tm}} dt}{\int I_{0.5\%\text{Tm}} dt} \quad (4)$$

where  $I$  denotes intensity, and  $x\%\text{Tm}$  represents the  $\text{Tm}^{3+}$  concentration. In fact, we have found that the results calculated from formulae (3) and (4) are very close and near.

Therefore, the energy transfer efficiencies of the 802.5, 648.0, and 456.0 nm fluorescences are calculated from formula (3) to be  $\eta_{\text{ET},3.8\%\text{Tm}}(802.5 \text{ nm}) = 65.5\%$ ,  $\eta_{\text{ET},3.8\%\text{Tm}}(648.0 \text{ nm}) = 59.5\%$ , and  $\eta_{\text{ET},3.8\%\text{Tm}}(456.0 \text{ nm}) = 39.7\%$ .

The schematic diagram of energy level structure and quantum cutting process is shown in Fig. 5. It can be seen from Fig. 1 that the intensity of the 802.5 nm excitation spectrum for (B)  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  powder phosphor is only slightly smaller than that of (A)  $\text{Tm}_{0.005}\text{Y}_{0.995}\text{NbO}_4$ . However, the intensity of the 1820.0 nm excitation spectrum for (B)  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  powder phosphor is much larger than that of (A)  $\text{Tm}_{0.005}\text{Y}_{0.995}\text{NbO}_4$ . The enhancement between the intensity of (B)  $\text{Tm}_{0.038}\text{Y}_{0.962}\text{NbO}_4$  and (A)  $\text{Tm}_{0.005}\text{Y}_{0.995}\text{NbO}_4$  for 1820.0 nm excitation spectrum is much larger than the reduction for that of 802.5 nm excitation spectrum. It illustrates that the luminescent energy of  $^3\text{H}_4$  level is transfer into  $^3\text{F}_4$  level. Moreover, it illustrates that it is the multi-photon process but linear process because one portion reduction in 802.5 nm luminescence intensity results in multi portions





CR9:ET<sup>r76</sup>-ET<sup>a01</sup>, and (8a) and (8b) for four-photon of CR10:ET<sup>r75</sup>-ET<sup>a03</sup> and CR12:ET<sup>r73</sup>-ET<sup>a04</sup> on best standard condition [1–2,5–8,12,14–17,19–20,25]:

$$\eta_{CR,x\%Tm}({}^3H_4) = \eta_{3H_4} \cdot [1 - \eta_{ET,x\%Tm}({}^3H_4)] + (1 - \eta_{3H_4}) \cdot \eta_{lower} \cdot [1 - \eta_{ET,x\%Tm}({}^3H_4)] + 2\eta_{3F_4} \eta_{ET,x\%Tm}({}^3H_4) \quad (5a)$$

$$\eta_{CR,x\%Tm}({}^3H_4)^{UL} = 1 + \eta_{ET,x\%Tm}({}^3H_4) \quad (5b)$$

In formula (5a) and (5b),  $\eta_{3F_4} \eta_{ET,x\%Tm}({}^3H_4)$  is the efficiency of population from  ${}^3H_4$  to  ${}^3F_4$  induced by cross-energy transfer.  $\eta_{3H_4} \cdot [1 - \eta_{ET,x\%Tm}({}^3H_4)]$  is the luminescence efficiency of the population remained in  ${}^3H_4$  state.  $(1 - \eta_{3H_4}) \cdot \eta_{lower} \cdot [1 - \eta_{ET,x\%Tm}({}^3H_4)]$  is the efficiency of population luminescent in lower level after they is from  ${}^3H_4$  to lower level induced by multi-phonon non-radiative relaxation.

$$\eta_{CR,x\%Tm}({}^1G_4) = \eta_{1G_4} \cdot [1 - \eta_{ET,x\%Tm}({}^1G_4)] + \{ (1 - \eta_{1G_4}) \cdot \eta_{lower} \cdot [1 - \eta_{ET,x\%Tm}({}^1G_4)] + \eta_{3H_4} \eta_{ET,x\%Tm}({}^1G_4) \} \cdot \{ \eta_{CR,x\%Tm}({}^3H_4) \} + \eta_{3F_4} \eta_{ET,x\%Tm}({}^1G_4) \quad (6a)$$

$$\eta_{CR,x\%Tm}({}^1G_4)^{UL} = 1 + \eta_{ET,x\%Tm}({}^1G_4) + \eta_{ET,x\%Tm}({}^1G_4) \cdot \eta_{ET,x\%Tm}({}^3H_4) \quad (6b)$$

$$\eta_{CR,x\%Tm}({}^1D_2) = \eta_{1D_2} \cdot [1 - \eta_{ET,x\%Tm}({}^1D_2)] + \{ (1 - \eta_{1D_2}) \cdot \eta_{lower} \cdot [1 - \eta_{ET,x\%Tm}({}^1D_2)] + \eta_{1G_4} \eta_{ET,x\%Tm}({}^1D_2) \} \cdot \{ \eta_{CR,x\%Tm}({}^1G_4) \} + \eta_{3F_4} \eta_{ET,x\%Tm}({}^1D_2) \quad (7a)$$

$$\eta_{CR,x\%Tm}({}^1D_2)^{UL} = 1 + \eta_{ET,x\%Tm}({}^1D_2) + \eta_{ET,x\%Tm}({}^1D_2) \cdot \eta_{ET,x\%Tm}({}^1G_4) + \eta_{ET,x\%Tm}({}^1D_2) \cdot \eta_{ET,x\%Tm}({}^1G_4) \cdot \eta_{ET,x\%Tm}({}^3H_4) \quad (7b)$$

$$\eta_{CR,x\%Tm}({}^1D_2) = \eta_{1D_2} \cdot [1 - \eta_{ET,x\%Tm}({}^1D_2)] + \{ (1 - \eta_{1D_2}) \cdot \eta_{lower} \cdot [1 - \eta_{ET,x\%Tm}({}^1D_2)] + \eta_{3H_4} \eta_{ET,x\%Tm}({}^1D_2) \} \cdot \{ \eta_{CR,x\%Tm}({}^3H_4) \} + \eta_{3H_4} \eta_{ET,x\%Tm}({}^1D_2) \cdot \{ \eta_{CR,x\%Tm}({}^3H_4) \} \quad (8a)$$

$$\eta_{CR,x\%Tm}({}^1D_2)^{UL} = 1 + \eta_{ET,x\%Tm}({}^1D_2) + 2 \cdot \eta_{ET,x\%Tm}({}^1D_2) \cdot \eta_{ET,x\%Tm}({}^3H_4) \quad (8b)$$

where  $\eta_{CR,x\%Tm}({}^3H_4)$  of (5a) and (5b),  $\eta_{CR,x\%Tm}({}^1G_4)$  of (6a) and (6b), and  $\eta_{CR,x\%Tm}({}^1D_2)$  of (7a)–(8b) are the two-photon, three-photon, and four-photon near-infrared quantum cutting efficiency of the

{CR1:ET<sup>r31</sup>-ET<sup>a01</sup>} of (5a) and (5b), {CR2:ET<sup>r63</sup>-ET<sup>a02</sup> and CR8:ET<sup>r62</sup>-ET<sup>a03</sup>} of (6a) and (6b), CR9:ET<sup>r76</sup>-ET<sup>a01</sup> of (7a) and (7b), and {CR10:ET<sup>r75</sup>-ET<sup>a03</sup>, CR12:ET<sup>r73</sup>-ET<sup>a04</sup>} cross-energy transfers of (8a) and (8b), respectively.  $\eta_{ET,x\%Tm}({}^3H_4)$ ,  $\eta_{ET,x\%Tm}({}^1G_4)$ , and  $\eta_{ET,x\%Tm}({}^1D_2)$  are the respective energy transfer efficiencies. And  $\eta_{1D_2}$ ,  $\eta_{1G_4}$ ,  $\eta_{3H_4}$ , and  $\eta_{3F_4}$  and  $\eta_{lower}$  are the luminescent efficiencies of the  ${}^1D_2$ ,  ${}^1G_4$ ,  ${}^3H_4$ , and  ${}^3F_4$  and lower energy levels of the  $Tm^{3+}$  ion, respectively. We assume that  $\eta_{1D_2} = \eta_{1G_4} = \eta_{3H_4} = \eta_{3F_4} = \eta_{lower} = 1$ , as is also assumed in most near-infrared quantum cutting literature [1–2,5–8,12,14–17,19–20,25]. Therefore, from (5a) and (5b),  $\eta_{CR,3.8\%Tm}({}^3H_4)^{UL} = \eta_{CR,3.8\%Tm}(682.5 \text{ nm})^{UL} = 165.5\%$ . Secondly, from (6a) and (6b)  $\eta_{CR,3.8\%Tm}({}^1G_4)^{UL} = \eta_{CR,3.8\%Tm}(461.0 \text{ nm})^{UL} = 198.5\%$ . From (7a) and (7b),  $\eta_{CR,3.8\%Tm}({}^1D_2)^{UL} = \eta_{CR,3.8\%Tm}(362.0 \text{ nm})^{UL} = 178.8\%$ . Finally, from (8a) and (8b),  $\eta_{CR,3.8\%Tm}({}^1D_2)^{UL} = \eta_{CR,3.8\%Tm}(362.0 \text{ nm})^{UL} = 191.7\%$ .

These analyses about the quantum cutting efficiency and mechanism are coincided with the experiment measured results. It is needed also to point out that the action of the phonon equal to 0 is not the most largest point for near infrared quantum cutting luminescence of  $Tm^{3+}$  ion. The most largest point for near infrared quantum cutting luminescence of  $Tm^{3+}$  ion would be positioned at the field which has a slight phonon action. Because quantum cutting needs higher doping concentration of rare earth ion, in order to achieve intense cross energy transfer and further to achieve intense near infrared quantum cutting luminescence. However, the actual material is difficult to possess the ideal condition of entire resonance for all energy transfer. It is excellent enough for that all energy transfers are all near entire resonant. Therefore it is needed a slight assistant action of phonon to achieve effective intense phonon-assistant energy transfers.

## 5. Conclusion

In conclusion, two-photon, three-photon, and four photon near-infrared quantum cutting luminescence of  $Tm^{3+}$  activator in  $YNbO_4$  powder phosphor is reported in present study. The two-photon, three-photon, and four-photon quantum cutting mechanism, induced by various energy transfer processes, are analyzed carefully. The largest up-limit of the two-, three-, and four-photons near-infrared quantum cutting efficiencies are found to be approximately 166%, 198%, and 192%, respectively.

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